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## IN THE SPECIFICATION

Please amend the following paragraphs in the specification to correct typographic errors.

[0023]. Syndiotactic styrene polymer (B) used in the invention is styrene polymer having high degree syndiotactic structure. As syndiotactic structure, where phenyl group or substituted phenyl group is positioned alternatively on the opposite side of side chain corresponding to the primary chain that is formed by carbon-carbon bond, and the tacticity is determined by Nuclear Magnetic Resonance method(13 C-NMR method). Nuclear Magnetic Resonance method(13 C-NMR method) determines the existing ratio of constitutional block of successive units, such as diode diad in case of 2 units, triode triad when 3 units and pentad for 5 units. In the invention, it is desirable to use styrene polymer having syndiotacticity at least 75%, preferably at least 85% in diode diad or at least 30%, preferably at least 50% in case of racemic pentad. The examples of styrene polymer are polystyrene, poly(alkylstyrene), poly (halogenated styrene), poly(alkoxystyrene), poly(vinylbenzoate ester) and their mixtures, or copolymer comprising mainly of these components. Furthermore, as poly(alkylstyrene), such as poly(methylstyrene), poly(ethylstyrene), poly(isopropylstyrene), poly(t-butylstyrene) etc.; poly(halogenated styrene), such as poly(chlorostyrene), poly(bromostyrene), poly(fluorostyrene) etc. are mentioned. As poly(alkoxystyrene), such as poly(methoxystyrene), poly(ethoxystyrene) etc. are listed.

[0025]. The preferred molecular weight of syndiotactic styrene polymer, although it is not restricted but generally weight-average molecular weight (Mw) is at least 10,000, and above all, the most appropriate is at least 50,000. If the weight-average molecular weight is less than 10,000, it has the tendency of weak chemical resistance. Furthermore, there are no restrictions about molecular weight distribution, and is appropriate for various things. The melting point of syndiotactic styrene polymer is in the range of 200 to 310°C, and has outstanding chemical resistance as compared to conventional styrene polymer having atactic structure.

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[0033]. As phosphate ester fire retardant, such as trimethyl phosphate, triethyl phosphate, trippentyl phosphate, tributyl phosphate, trippentyl phosphate, trihexyl phosphate, tricyclohexyl phosphate, trippentyl phosphate, trippen

[Formula 2]

$$Ar_1 = O = P = O = P = O = Ar_3$$

$$Ar_2 = Ar_4 = R$$

式中、Rは下記式A1~A4より選ばれる基

$$Ar^{1} \longrightarrow O \longrightarrow P \longrightarrow O \longrightarrow R \longrightarrow O \longrightarrow P \longrightarrow O \longrightarrow Ar^{3}$$

$$Ar^{2} \longrightarrow Ar^{2} \longrightarrow Ar^{4} \longrightarrow m$$

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wherein, R is the group selected from A1 to A4; wherein, n varies from 1 to 10, Ar<sub>1</sub> to AR<sub>4</sub> are phenyl group, tolyl group or xylyl group respectively. Further, when n is at least 2, multiple AR<sub>4</sub> may be same or different). From the viewpoint of fire retardancy and heat-resistant, it is desirable that R is the structure of (A4). These can be used independently or at least two types can also be used together. The preferred phosphate ester fire retardant is aromatic phosphate ester having aromatic group.